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When writing the chemical formula for an ion, its net charge is written in superscript immediately after the chemical structure for the molecule/atom. The net charge is written with the magnitude before the sign; that is, a doubly charged cation is indicated as 2+ instead of +2.However, the magnitude of the charge is omitted for singly charged molecules/atoms; for example, the sodium cation is ...

Ion - Wikipedia

Inorganic chemistry deals with synthesis and behavior of inorganic and organometallic compounds. This field covers chemical compounds that are not carbon-based, which are the subjects of organic chemistry.The distinction between the two disciplines is far from absolute, as there is much overlap in the subdiscipline of organometallic chemistry.It has applications in every aspect of the chemical ...

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