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neutron in the nucleus of uranium: A. 94  
B. 145 C. 235 D. Same number E.  
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radon specimen emits radiation of  $3.7 \times 10^3$  GBq per second.

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Nucleus consists of: Z protons with e+ charge  
N neutrons with no charge. A  
Mass number  $A=Z+N$  protons &

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neutrons are bound by strong force: R  
10-13 m. Nomenclature and common  
units Power Prefix Abbrev. 10-15 femto f  
10-12 pico p 10-9 nano n 10-6 micro  $\mu$   
10-3 milli m 10-2 centi c 10-1 deci d 10<sup>3</sup>  
kilo k

## **1.3. Basic Principles of Nuclear Physics**

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[ $^{14}\text{N}(n,p)\text{C}$ , the neutrons being produced by cosmic rays].  $^{14}\text{C}$  decays by  $\beta$ -emission with half life  $T_{1/2} = 5730$  y  $^{14}\text{C} \rightarrow ^{14}\text{N} + \beta^- + \bar{\nu}$  The chemical activity of  $^{14}\text{C}$  is similar to that of  $^{12}\text{C}$ , so living organisms have the same  $^{14}\text{C}/^{12}\text{C}$  ratio as in the atmosphere, which is about  $1.35 \times 10^{-12}$ . When an organism dies it stops absorbing  $^{14}\text{C}$ , and the

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ratio

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